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Excess volumes (x < 0.1 or x > 0.9) and excess enthalpies (x > 0.9) of the mixtures  $\{x \ 1,4\text{-dioxane or } 1,3\text{-dioxane} + (1-x) \text{ non-polar species}\}$  at 298.15 K : the partial molar excess volumes and the partial molar enthalpies.\*

## by

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## Abstract

Excess enthalpies  $H^{E}$  and excess volumes  $V^{E}$  of the mixtures of  $\{x \ 1,3\text{-dioxane or } 1,4\text{-dioxane}+(1-x) \ 2,2,4\text{-trimethylpentane}$ , or benzene, or tetrachloroethene $\}$  (x > 0.9) were measured at 298.15 K, and excess volumes of the region in x < 0.1 for those mixtures were also measured. Partial molar excess enthalpies  $H^{E}_{i}$  and partial molar excess volumes  $V^{E}_{i}$  of those mixtures were estimated from them, and those at infinite dilution  $H^{E*}_{i}$  and  $V^{E*}_{i}$  were discussed, together with partial molar excess enthalpies of 1,3-dioxane and 1,4-dioxane at infinite dilution in the region (x < 0.1) of those mixtures were also reported.

It is revealed that the local dipole moment of 1,4-dioxane plays an important role and dominantes in the intermolecular interactions in the mixtures, as that of 1,3-dioxane.

Keyword : Partial molar excess enthalpy; Partial molar excess volume; 1,3-Dioxane; 1,4-Dioxane; Non-polar solvents.

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